



#### PDR Theory: Overview Michael Kaufman - SJSU Physics & NASA Ames

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D. Neufeld (JHU) R. Plume (U Calgary) Dense Gas  $n \sim 10^4 \text{ cm}^{-3}$ 

# Overhead View $n \sim 10^1 - 10^2 \text{ cm}^{-3}$

Diffuse Gas

FIR Emission from the ISM Dominated by PDRs → FUV controls heating/chemistry to A<sub>v</sub> ~ 5-10 in neutral gas

- Neutral atomic diffuse/translucent clouds ( $A_V \le 1-2$ )
- Surfaces of Molecular clouds
   H → H<sub>2</sub> transition (A<sub>v</sub> ~ 1- few)
- Clumpy GMCs FUV permeates clouds

"All the gas in a galaxy is in a PDR!"

# Model Inputs

Geometry

#### UV Field and Penetration

Grains/PAHs

 $H_2$  Formation

## Chemistry

## Line transfer/cooling

**Cosmic Ray Ionization** 

Photoelectric Heating

## **PDR Schematic**





# Outline

- Radiative transfer
  - FUV
  - dust emission
- Role of Molecular Hydrogen
  - formation
  - detailed treatment of line emission
- Heating & Cooling
- Chemistry

## **FUV** Penetration

- Continuum attenuated by dust
- Depends on absorption/scattering properties measured in diffuse clouds (e.g. Draine & Lee 84)
- Roberge et al. (1991) and van Dishoeck et al (1988) calculated unshielded photodissociation/photoionization rates in local ISRF, fit decrease in rates with increasing A<sub>V</sub>... good for R=3.1! (N.B. Dense clouds have higher R values)



## Spherical Geometry



## Dust Emission

- Usually not the goal of PDR models → very approximate treatments
- e.g.: calculate  $T_d$  at surface and  $\tau$  to reemitted surface IR at all other depths
- Observations (esp. extragalactic) would benefit from more accurate modeling → get G<sub>0</sub> from the SED

## Molecular Hydrogen

Key to structure of PDRs

Dominant species (once it forms)
Kicks off all molecular chemistry
Heats gas under certain conditions
Formation pumping can drive chemistry
=>Get it right!

# H<sub>2</sub> Formation

- Gould & Salpeter (1963) recognized inefficiency of gas-phase H<sub>2</sub> formation → suggested grains as catalytic sites
- Hollenbach & Salpeter (1971) : H must be chemisorbed to allow efficient formation at high grain temperature
- Numerous attempts at laboratory, theoretical, and observations constraints on formation rate ==> KEY to PDR structure

## Amongst the rates are....

- $R = 3x10^{-18} f S T^{1/2} cm^3 s^{-1}$  where f = f( $T_d, E_b$ ),  $S = S(T, T_d)$  [Hollenbach & Salpeter 71, Sternberg & Dalgarno 95]
- Same, but with assumption that f,S=1
- $3 \times 10^{-17} \text{ cm}^3 \text{ s}^{-1}$  under assumption that  $\overline{T}_{gas} \sim 100 \text{ K}$  in formation region and other properties are too poorly known anyway!

#### H<sub>2</sub> formed but grains too cold ==> sites blocked

#### Thermal hopping allows H to find chemisorbed H

# Warm grains evaporate H before recombination



# Observational Constraints on H<sub>2</sub> Formation





Habart et al. 2004

 $R \sim 3x10^{-17} - 2x10^{-16} \text{ cm}^3 \text{ s}^{-1}$ 



# H<sub>2</sub> Pumping/dissociation/heating



•FUV pumping of H<sub>2</sub> in Lyman/Werner bands (11.3-13.6 eV)• ~90% fluoresce to to bound vibrational state with  $\sim 2eV \implies$  emitted in IR lines (low n) **or** heat gas by collisional deexcitation (high n)  $\sim 10\%$  fluoresce to vibrational continuum ==> dissociation Very detailed treatment: Le Bourlot et al. 1993, ==> full accounting Simplified : deexcitation into "pseudo-level" ==>OK for approximate heating, bad for specific line strengths

## Gas Heating & Cooling

- Heating

   Grain photoelectric
  - H<sub>2</sub> formation
  - cosmic rays
  - grain drift
- Cooling
  - line emission

## Heating Processes



## **Photoelectric Heating**

a ~ 4 - 150 Å

Dust absorbs most of the FUV:  $\sigma_{dust} > \sigma_{gas}$ 

FUV

99%: photon absorbed by dust  $\rightarrow$  warm dust

1% : "photoelectric ejection"  $\rightarrow$  warm gas

Collides w/ gas  $\rightarrow$ gas heating  $\rightarrow$  line cooling

IR Continuum

 $\rightarrow$  gas cooling/dust cooling = line/continuum ~ 1%

## **Photoelectric Heating**



## Cosmic Rays



## Chemistry

- By definition, influenced by FUV photons (unlike dense cores where CR's dominate)
- Photo-reactions
- H<sub>2</sub> formation kicks off all other molecular species
- C<sup>+</sup> and O maintained to high depths
- Vibrationally-excited H<sub>2</sub>

## The Chemical Effects of PAHs



 $PAH + e \implies PAH^{-}$   $PAH^{-} + C^{+} \implies PAH + C$ Result: N(C) up by 2,  $N(C^{+}) down by 2$   $I(C)/I(C^{+}) up by 4$ 

## Example: Rates for OH + O



## Example: H<sub>2</sub>O in PDRs

- Formation:
  - $\begin{array}{ll} H_{3}O^{+} + e^{-} \rightarrow H_{2}O + H & (0.05 \text{ or } 0.25) \\ \rightarrow & OH + 2H & (0.65 \text{ or } 0.75) \\ \rightarrow & O + H_{2} + H & (0.30 \text{ or } 0.00) \end{array}$
- Destruction  $H_2O + FUV \text{ photon} \rightarrow OH + H$  $C^+ + H_2O \rightarrow HCO^+ + H$

# W51 $H_2O$ Absorption: First Detection of $H_2O$ in Diffuse Gas







# Modeling

